

## THE EFFECTS OF ADSORBENT MATERIALS ON THE LIPID QUALITY OF LEMURU FISH OIL AND THE ENRICHMENT OF OMEGA-3 USING LIPASE

[Pengaruh Material Adsorben terhadap Kualitas Lipida Minyak Ikan Lemuru dan Pengayaan Omega-3 Menggunakan Lipase]

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### ABSTRACT

Sardinella (lemuru) is a genus of fish that is widespread in the East Java Sea region, especially in Muncar, Banyuwangi in East Java Province, Indonesia. Marine fishes are rich in essential fatty acids, including omega-3. This study aims to improve the quality of lemuru fish oil through the use of adsorbent materials such as bentonite and activated carbon. The sample was analyzed to determine its free fatty acid (FFA) content and peroxide value (PV) using the titrimetric method. The clarity of the oil was determined using the spectrophotometric method ( $\lambda$  440 nm). The omega 3 content from the fish oil sample was then enriched through enzymatic reactions using lipase. Enzymatic reactions were carried out for 5, 10, 15, 20, 35, and 47 hours with a lipase concentration of 500, 1000, 1500, and 2000 units respectively. The omega-3 content of the fish oil products was analyzed using the GC-FID method. The results show that the addition of 3% activated carbon and bentonite in the fish oil reduced the FFA and PV results. Moreover, the absorbance value at  $\lambda$  440 nm was also reduced from 0.883 to 0.559. The highest content of omega-3 was obtained through hydrolysis with lipase at a concentration of 1000 units for 35 hours with alpha-methyl linolenate (ALA), methyl all-cis-5,8,11,14,17-eicosapentanoate (EPA) and cis-4,7,10,13,16,19-docosahexaenoic acid methyl ester (DHA) at 0.78, 1.06, and 0.29% respectively.

**Keywords:** activated carbon, adsorbent, bentonite, lemuru fish oil, purification

### ABSTRAK

Ikan sardinella (lemuru) adalah salah satu jenis ikan yang tersebar luas di wilayah Laut Jawa Timur, terutama di Muncar, Banyuwangi, Provinsi Jawa Timur, Indonesia. Ikan laut kaya akan asam lemak esensial seperti omega-3. Penelitian ini bertujuan untuk meningkatkan kualitas minyak ikan lemuru dengan menggunakan bahan adsorben berupa bentonit dan karbon aktif. Minyak ikan lemuru dimurnikan dengan bentonit dan karbon aktif. Sampel minyak ikan kemudian ditentukan nilai asam lemak bebas dan angka peroksidanya dengan metode titrimetri, sedangkan kejernihan minyak ditentukan dengan metode spektrofotometri pada  $\lambda$  440 nm. Kandungan omega-3 minyak ikan diperkaya melalui reaksi enzimatik menggunakan enzim lipase komersial. Reaksi enzimatik dilakukan selama 5, 10, 15, 20, 35, dan 47 jam dan konsentrasi enzim lipase berturut-turut 500, 1000, 1500, dan 2000 unit. Kandungan omega-3 dari produk minyak ikan dianalisis menggunakan kromatografi gas dengan detektor FID. Hasil penelitian menunjukkan bahwa penambahan 3% karbon aktif dan 3% bentonit pada minyak ikan dapat menurunkan nilai angka peroksida dan asam lemak bebas. Selain itu nilai absorbansi pada  $\lambda$  440 nm juga berkurang dari 0,883 menjadi 0,559. Kadar omega-3 tertinggi diperoleh melalui hidrolisis dengan lipase pada konsentrasi 1000 unit selama 35 jam dengan kadar alpha-methyl linolenate (ALA), methyl all-cis-5,8,11,14,17-eicosapentanoate (EPA), dan cis-4,7,10,13,16,19-docosahexaenoic acid methyl ester (DHA) masing-masing sebesar 0,78; 1,06; dan 0,29%.

**Kata kunci:** bentonit, karbon aktif, minyak ikan lemuru, pemurnian, reaksi enzimatik

## INTRODUCTION

Fatty acids and fish oil have become a prominent area of interest because of omega-3 (Hong *et al.*, 2015). Fish oil originates from the oil-rich fish tissue that contains omega-3 fatty acids (Abu-Ouf and Jan, 2014). The dominant types of omega-3 in fish oil are EPA and DHA (Suriani *et al.*, 2019; Aryani *et al.*, 2017). Previously, fish oil was a product of the fish meal used for animal feed. This is now recognized as the primary source of these fatty acids (Bonilla and Concha, 2018). Among the sources in the human diet, fish has been highly regarded as an easy access yet highly nutritious food due to its high protein content and the presence of polyunsaturated fatty acids (PUFA), particularly, omega-3 fatty acid, eicosapentaenoic acid and docosahexaenoic acid (Panagan *et al.*, 2012; Raharja, 2012; Bija *et al.*, 2016; Ahmed *et al.*, 2017; Akbar *et al.*, 2017; Maki *et al.*, 2017; Srigley and Rader, 2014; Moharana *et al.*, 2016; Eltweri *et al.*, 2017; Yang *et al.*, 2020). Raw marine fish oil extracts also contain five new arsenolipids (Pereira *et al.*, 2016). The remarkable impact of omega-3 fatty acids is closely linked with healthy aging, as well as other substantial health effects (Kolanowski, 2010). Other potencies include the omega-3 fatty acid potential to exhibit pleiotropic cardiometabolic effects via multiple actions (Kromhout, 2012), contributing to gestational diabetes mellitus effects (Ostadrahimi *et al.*, 2016) and preventing cardiovascular disease and cancer (Billman and Harris, 2011; Mozaffarian and Wu, 2011; Schuchardt *et al.*, 2011; Wen *et al.*, 2014; Dinicolantonio *et al.*, 2014; Mori, 2014; Calvo *et al.*, 2017; Hu *et al.*, 2019; Manson *et al.*, 2018; Fonda *et al.*, 2016; Minihane, 2013; Pirillo and Catapano, 2013; Siscovick *et al.*, 2017; Stephenson *et al.*, 2013; Mohebi and Bikdeli, 2014; Yinko *et al.*, 2014; Nestel *et al.*, 2015; Endo and Arita, 2016; Harris *et al.*, 2018; Jump *et al.*, 2012; Ferrari *et al.*, 2020; Lombardi *et al.*, 2021; Jo *et al.*, 2021; Elagizi *et al.*, 2021; Weinberg *et al.*, 2021).

The addition of fish oil can cause problems, specifically in terms of the odor and taste of most food products. The odor and taste of fish oil is what consumers dislike the most. However, modern processing technology can reduce these problems through purification techniques. The free fatty acids can be removed through deacidification (neutralization), while the color and oxidation can be removed through bleaching. The flavor can be removed through the process of deodorization, The PUFA concentration is increased through winterization and/or molecular distillation, and oxidation can be prevented by adding antioxidants. Bleaching is a noteworthy method in oil refinement (Monte *et al.*, 2015; Kralovec *et al.*, 2012). Color properties have a significant effect on consumer acceptability,

especially edible products, beside its nutritional value and sensory properties. Upon the consumer's consideration, the color of the food directly shows its quality and maturity (for fruits and vegetables), which links to the customer expectations, their like or dislike of it according to their preference, and its acceptability. The color of the product is also an important consideration among technologists. The color properties indirectly show the consistency, stability and composition of the product, which also lead to solving the presented problem by enhancing its appeal and appearance to meet the customer needs (Francis, 1999).

The oil purification process is beneficial as it removes impurities, lowers the free fatty acid content, and clears the oil color. Bleaching is the process used to refine the oil color because fish oil derived from waste tends to not look appealing. It needs to be refined through bleaching, which utilizes adsorbent.

Unrefined fish oil is still used as an animal feed ingredient in Indonesia. It contains soap stock, and high primary and secondary lipid oxidation product content which have been the problems associated with color, odor, flavor, and other impurities. Fish oil refining and purification through an effective and efficient method is very important for improving the quality of fish oil that is suitable for consumption.

An adsorbent can adsorb the impurities of the components, pigments, and free fatty acids in the fish oil. Adsorbents, which are potentially used in purification process, include activated carbon, attapulgite, bentonite, and chitosan (Suseno *et al.*, 2013). The amount of adsorbent used in the bleaching process varies depending on the type of oil, the intensity of the oil color, and the desired color of the bleached oil (García-Moreno *et al.*, 2013). Adsorbent refining is a common technique used to enhance the quality of edible oil. Adsorbents are usually used in the bleaching process to remove the color pigment, which is an undesirable compound in the oil (Ketaren, 2012).

The main types of pelagic fish are lemuru fish, which are commonly found in the waters of the Bali Strait (Dari *et al.*, 2017). Sardine fish or lemuru fish are one of the potential sources of a high level of lipids that contain the crucial EPA and DHA. Sardine fish processing waste includes the liver, head, and intestines. The results of the analysis of oil extraction from the processing of these ingredients indicates that there is an essential PUFA content (Kones and Rumana, 2017).

The lemuru fish treatment by product characteristic is that it exhibits a fluctuating amount of saturated fatty acids. According to the previous study, the treatment uses a variety of bleach components such as 5% activated charcoal and zeolite, and 5% bentonite (Saraswati, 2013). A bentonite

concentration of 5% is effective for improving the physical and chemical quality of rice bran oil. The bleaching process utilizes 25% HCl activated zeolite. The bleaching material, in the form of clay, is activated at 90°C at 5% of the weight of oil for 60 min (García-Moreno *et al.*, 2013). This process is able to improve the quality of the fish oil by products following the lemuru milling process.

On an industrial scale, conventional oil purification processes are usually carried out through several different chemical methods including the process of separating the phospholipids through a degumming process using a 5% magnesol XL adsorbent (Haryati *et al.*, 2017), deacidification using conventional solvent extraction and membrane deacidification, the bleaching process in various conditions (Charanyaa *et al.*, 2017), and the addition of odor removal agents to reduce the compounds that cause the odor. However, there are some weaknesses regarding these processes including the occurrence of environmental pollution due to the use of alkali chemicals and the loss of some neutral oil. To eliminate the volatile compounds and free fatty acids in the physical purification of the oil, an alternative method is proposed by applying high-temperature steam at a low pressure. However, in practice, this process still requires an initial refining process that uses chemicals. This is not suitable for oils that are vulnerable to high temperatures such as fish oil (Čmolík and Pokorný, 2000). To remove contaminants in fish oil such as dioxins and PCBs, recently the use of activated carbon has been proposed, involving a physical adsorption process (Maes *et al.*, 2005). The odor of raw fish oil is highly disturbing and limits its application in the food industry. Thus, the process of removing the odor becomes an important step. Deodorizing traditional oils can be done by heating them at high temperatures. In this study, fish oil refinement was carried out using a combination of two adsorbents, specifically activated carbon and bentonite.

In this study, the purification of fish oil was carried out using carbon black and bentonite as trial adsorbents because they are easy to find in daily life compared with another adsorbent, magnesol XL. In this study, it is expected that we will be able to determine the right type of adsorbent and concentration to improve the quality of lemuru fish oil. This study aims to improve the quality of lemuru fish oil through adsorption by decreasing the amount of free fatty acids (FFAs) to reduce oxidation. Oxidized products can cause rancidity in fish oil. Decreasing the peroxide value (PV) avoids damage to the fish oil and removes the foul odor from the oil. The clarity of the fish oil goes slightly yellow, enriching the omega-3 content through an enzymatic reaction using lipase.

## MATERIALS AND METHODS

### Materials

The materials used in this study included lipase enzyme powder (200,000 unit/g) purchased from Xi'an Lyphar Biotech Company (Shaanxi, China). The phosphate buffer, boron trifluoride (BF<sub>3</sub>), ethanol, KOH, phenolphthalein, sodium thiosulfate, acetic acid, chloroform, methanol, potassium iodide, and n-hexane (Merck, USA). The omega-3 standard, alpha-methyl linolenate (ALA) (Merck, USA) and methyl all-cis-5,8,11,14,17-eicosapentanoate (EPA), and cis-4,7,10,13,16,19-docosahexaenoic acid methyl ester (DHA) (Sigma Aldrich, USA).

### The absorption treatments of the lemuru fish oil

The sample used in this study was lemuru fish oil from a local fish processing unit manufacturer, CV. Biji Sesawi (Banyuwangi, East Java Provinces, Indonesia). The lemuru fish oil was placed in a container and stored for about 10-30 day at 4°C. The absorption treatments of the lemuru fish oil were based on the research by Nadia *et al.* (2020) and Ayu *et al.* (2020) with some modifications. The lemuru fish oil was prepared in 100 mL volumes for each treatment. The treatments were carried out by adding 1, 2, and 3% activated carbon or bentonite as the adsorbent with 2 replications before being heated at 800°C for 30 minutes. The lemuru fish oil samples were then stirred at room temperature and centrifuged at 6000 rpm for 10 minutes. The first layer was put into a vial for further analysis.

### Determination of free fatty acids (FFAs) (AOAC, 2005)

Lemuru fish oil (10 g) was added to 50 mL of 95% ethanol and boiled for 10 minutes. Two drops of 2% phenolphthalein were added to mixture and then titrated with KOH 0.1 N until a pink color appeared. FFA was expressed as a percentage of 0.862%. This stage of the analysis was replicated twice for each bentonite concentration. The results of the FFA analysis obtained a result of (0.862±0.021)%.

### Determination of the peroxide value (PV) (AOAC, 2005)

The lemuru fish oil (5 g) was put into a 250 mL Erlenmeyer flask. A volume of 30 mL of a mixture of acetic acid and chloroform was added with a ratio of 3:2, and then 0.5 mL of potassium iodide was added and the solution was mixed before the addition of 30 mL of distilled water. The solution was titrated with 0.01 N sodium thiosulfate until it turned yellow. Subsequently, 0.5 mL of 1% indicator starch solution was added until a blue color appeared in the solution. The titration process was continued by shaking the solution until the color of the solution

turned light blue, indicating the release of iodine from the chloroform layer. Titration was performed until the disappearance of the blue color from the solution. The PV was expressed as meq/Kg. This stage of the analysis was replicated twice for each activated carbon concentration. The results of the FFA analysis obtained a value of 0 meq/Kg.

### Enzymatic hydrolysis of the fish oil

The omega-3 content of the treated fish oil was enriched through an enzymatic reaction using lipases. Briefly, the lemuru fish oil (0.6 g) was mixed with 1.3 mL hexane, lipase (1000 u/mL) and 0.1 M phosphate buffer pH 5.7 until 3 mL total volume was reached. The solution was incubated at room temperature and shaken at 150 rpm. The reaction was ended through the addition of 2 mL methanol. The enzymatic hydrolysis was performed at 5, 10, 15, 20, 35, and 47 hours.

### Preparation of the omega-3 standard solution and methylation (AOAC, 2005)

The stock solutions of ALA, EPA and DHA were prepared by dissolving each substance in hexane at 50 mg/mL. The solutions were subsequently stirred to prepare the mixture of standard solution at a concentration of 5000 µg/mL. For the linearity determination, the standard mixture solution was diluted to a concentration range of 250 to 2000 µg/mL in hexane. For the methylation, the fish oil (0.125 g) was placed in a flask followed by the addition of 0.5 mL BF<sub>3</sub> in MeOH (14%). BF<sub>3</sub> was used due to a better chromatographic outcome in a preliminary experiment compared to other solvents such as potassium hydroxide and sodium methoxide. The flask containing the fish oil and BF<sub>3</sub> was incubated in an incubator shaker at 55°C for 1.5 hours. Afterwards, 0.5 mL saturated NaCHO<sub>3</sub> and 1.0 mL hexane were added to the flask. The mixture was mixed and shaken well using a vortex for 30 seconds. The mixture was placed in a freezer for 10 min to obtain a two-layer formation. A volume of 0.5 mL hexane from the upper layer was carefully transferred into a vial for gas chromatography (GC) analysis.

### Omega-3 fatty acid analysis (AOAC, 2005)

The fatty acid composition of the fish hydrolysate samples was analyzed using GC Agilent 7890B (Agilent Technologies Inc., Santa Clara, USA) equipped with a split injector and a flame ionization detection (FID) system in order to quantify each fatty acid methyl ester (FAME) component. The FAMES were separated using the HP-5 column instrument (30 mx0.32 mm i.d, 0.25 µm). The oven temperature was held stable at 100°C for 2 minutes and increased to 240°C at 10°C/minutes, then held for 1 minutes. The temperatures of the detector and

injector were kept at 300 and 250°C respectively. The sample (1 µL) was injected with a split ratio of 100:1. The carrier gas used by the system was helium amounting to 3.0 mL/minutes and controlled at 15.726 psi. The hydrogen and air used by the FID were maintained at 30 and 400 mL/minutes respectively. All omega-3 peaks were identified based on their retention time according to the corresponding standards. The omega-3 concentration was determined by plotting the peak area against the calibration curve of each compound (ALA, EPA and DHA). The omega-3 fatty acid concentration was expressed as ppm. The analysis was replicated twice. The following results of the FFA analysis were obtained: (869.58±0.721) ppm or (0.870±0.00007)%.

## RESULTS AND DISCUSSION

### Free fatty acids (FFAs)

The less attractive taste in oil is related to the content of FFAs. The value of the FFAs in the fish oil processing industry is primarily determined by the amount of alkali used in the refining process. FFAs are an indication of the hydrolytic rancidity of the FFA levels determined by standard alkaline titration (Azman *et al.*, 2018). FFAs are a product of the triacylglyceride hydrolysis reaction and are closely related to the storage process (Nurjanah *et al.*, 2014). Factors that can affect the value of free fatty acids include poor storage, which can increase the fatty acid levels. Fatty acid oxidation is highly dependent on the number of double bonds and is influenced by temperature, oxygen concentration, metals, water activity, prooxidants, antioxidants, and catalysts. PUFA components, which are mostly found in fish oil, have a number of long double bonds that are easily oxidized compared to black seed oil, which is composed of mono-unsaturated fatty acids. The increase in the value of the FFAs in the oil can increase oxidation. An oxidized product can result in the rancidity of the aforementioned oil. Ghani (2014) reported that adsorbents function as a substance used to absorb the components of the FFAs within the oil without hydrolyzing the oil itself. The results of the FFA test are presented in Figure 1.

The FFAs were evaluated by comparing the lemuru fish oil without and with the addition of 1, 2 and 3% activated carbon or bentonite. The FFA testing varied between 0.862 and 1.2098%, which showed a high accuracy compared to the FFA testing of 1.68% found in the previous study (Nadia *et al.*, 2020). The standard deviation was between 0.021 and 0.043%. The limit required for the FFAs by the International Fish Oil Standard (IFOS) 2014 is FFA≤1.5% (IFOS, 2014). An FFA-focused study of lemuru fish oil can indicate the tolerance values

within the allowable limit. Figure 1 demonstrates that bentonite results in a more effective reduction of FFAs than activated carbon. This is because the ability of the activated carbon to adsorb FFAs is smaller than that of bentonite. The smaller adsorption ability is probably due to the larger size of the activated carbon particles compared to bentonite. One of the factors that influences adsorption is the size of the adsorbent particles. The smaller the particle size, the greater the diffusion rate of the solute molecules into the adsorbent pores (Rio *et al.*, 2009). The activated carbon particles are granular, whereas bentonite particles are powder.

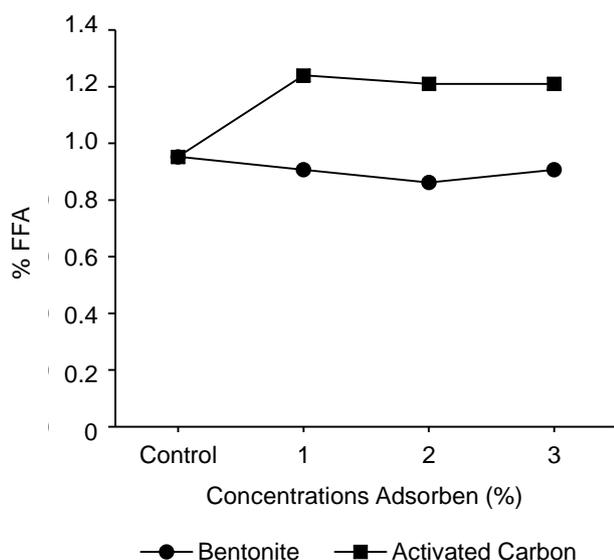


Figure 1. The FFA of the lemuru fish oil treated with bentonite and activated carbon

### Peroxide value (PV)

Fish oil quality level is determined by the PV. Peroxide is formed by the binding of oxygen into the double bonds found in unsaturated fatty acids. Gunawan *et al.* (2003) explained that oil that is in direct contact with air and high temperatures can result in the unsaturated fatty acids breaking down. The carbon chain in the double bond is broken, resulting in a free fatty acid and peroxide increase. Inadequate processing, handling, and an unstable extraction temperature of the fish oil results in a high PV, as well as accelerating oxidation. The smaller the PV, the better the quality of the oil (Panagan *et al.*, 2011). Factors that can cause a high PV include high temperatures and the type of clear packaging that will accelerate the oxidation process in the oil. Peroxide values that exceed the standard can cause the body to be poisoned if consumed (Ketaren, 2012). It accelerates the smell of rancidity and has an unliked flavor, and it is toxic to the body if the peroxide value is more than 100 meq/Kg (Nurhasnawati *et al.*, 2015). The increase of the

peroxide number indicates a rise in peroxide that causes damage to the oil and gives off foul odor.

Before adding activated carbon, the peroxide value of the lemuru fish oil was determined to be 42 meq/Kg as shown in Figure 2. The peroxide value was high while the limit required for the peroxide value by the IFOS (2014) is  $PV \leq 5$  meq/Kg (IFOS, 2014). The oxidation process of the oxygen in the air can be the contributing factor in the alteration of the unsaturated fatty acids, as it decays the oil or fat (Panagan *et al.*, 2011). Unsaturated fatty acids are significantly reactive to oxygen, increasing the number of double bonds in the molecular chain. The spontaneous oxidation of bland fatty acids is based on the intervention of oxygen in the double bonds to form peroxides (Panagan *et al.*, 2011).

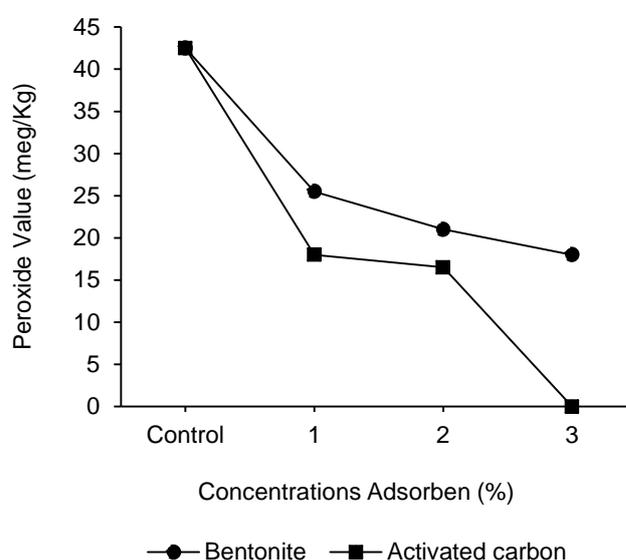


Figure 2. Peroxide value of the lemuru fish oil treated with bentonite and activated carbon

The PV was evaluated by comparing the lemuru fish oil with and without the addition of 1, 2 and 3% activated carbon and bentonite. As shown in Figure 2, the addition of activated carbon to the lemuru fish oil is more effective than bentonite at decreasing the PV because activated carbon has a large surface area and pores that allow it to bind and adsorb more peroxide compounds in the oil (Miskah *et al.*, 2019). This result indicates that activated carbon played a significant role in decreasing the PV from 42.5 to 0 meq/Kg. This was highly accurate compared to the PV testing of 58.38 meq/Kg found in the previous study (Nadia *et al.*, 2020).

The initial characterization data of the lemuru fish oil sample showed that the concentration of FFA and the peroxide number were under the safe limits to be consumed based on commercial fish oil standard by the IFOS. This means that the purifica-

tion process carried out played a significant role in reducing the FFA content and PV.

### Clarity

The clarity level of the fish oil was presented in the form of light absorbance as shown on the spectrophotometer. Measurements were made at a wavelength of 440 nm. The highest clarity level obtained from 3% activated carbon was able to reduce the absorbance from 0.883 to 0.559 as shown in Figure 3. This is close to the clarity of commercial fish oils. Through this treatment, the color of lemuru fish oil changed from opaque to slightly yellow.

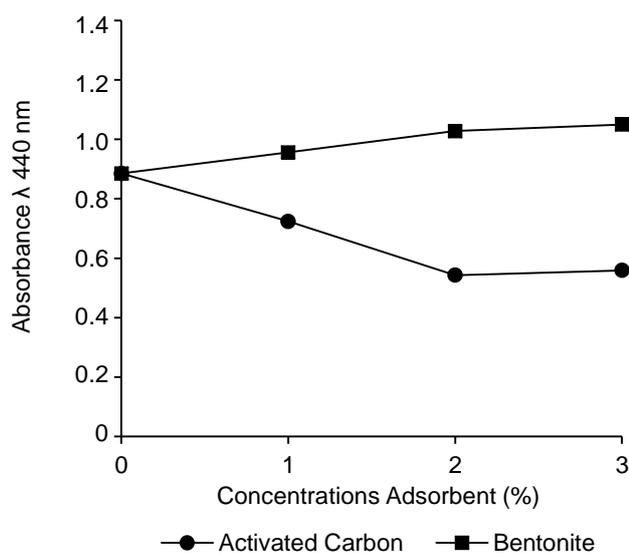


Figure 3. Clarity of the lemuru fish oil

Activated carbon played a significant role due to absorbing the color of the fish oil through its pores. Primary and secondary oxidation products tend to affect the color and turbidity of fish oil. The higher the amount of primary and secondary oxidation products, the darker the color, and the more the clarity is decreased. The presence of impurities such as slime, sap and gum in fish oil will result in a higher level of absorbance.

### Omega-3 analysis

Pertaining to the saturation of carbon, the separation of the fatty acids can be accomplished using adsorbents. HPLC and silver resin chromatography have been employed for the preparation of

omega-3 concentrates (Shahidi and Wanasundara, 1998). Analytical gas-liquid chromatography enables the excellent separation of polyunsaturated fatty acid esters but it is difficult to scale up. In this study, the omega-3 fatty acid analysis was performed using GC-FID. All peaks provided an excellent resolution, as shown by the distinct separation of the base peaks. In relation to this result, the analysis of the omega-3 was performed using an HP-5 capillary column paired with the GC-FID.

The linearity of the omega-3 determination using GC-FID was investigated, and the results are presented in Table 1. For all omega-3 tests, the linear range was 0 to 2000  $\mu\text{g/mL}$  at a correlation coefficient of  $\geq 0.999$ . Figure 4 shows the chromatogram of lemuru fish oil following the hydrolysis process using lipase, which results in the presence of  $\alpha$ -linolenic acid, eicosapentaenoic acid and docosahexaenoic acid. The chromatogram reveals that the HP-5 capillary column was selective when it came to separating out the components in the lemuru fish oil.

### The effect of enzyme concentration on the omega-3 content

The omega-3 content obtained from fish oil treatments was enriched through an enzymatic reaction using lipase to improve the quality of the lemuru fish oil. The enzyme concentration and incubation time were evaluated. The enzymatic hydrolysis of lemuru fish oil was carried out by evaluating the effects of enzyme concentration and the incubation process time on the omega-3 content. The optimum enzyme concentration in the enzymatic hydrolysis process is shown in Figure 5. The increase in enzyme concentration accelerated the reaction with the function of getting the available substrates to bind. After all of the substrates were bound, the reaction was no longer accelerated because there was no more binding to the additional enzymes. The optimum enzyme concentration was found by varying the enzyme concentration to produce a high omega-3 content. Figure 5 presents the optimal amount of enzyme for 1000 units/g of fish oil with which to obtain ALA, EPA, and DHA with concentrations of 0.870, 2.56, and 0.30% respectively. This was a more efficient analysis compared to the enzyme concentration of 20,000 units/g found in the previous study (Nugrahini *et al.*, 2016).

Table 1. Calibration curve data of the omega-3 fatty acid standards

Compound	tR (min)	Linear Range ( $\mu\text{g/mL}$ )	Calibration Data	
			Correlation Coefficient (R)	Equation
ALA	14.001	0-2000	0.999	$y=1.102+0.142x$
EPA	15.431	0-2000	0.999	$y=0.391+0.036x$
DHA	16.947	0-2000	0.999	$y=1.628+0.139x$

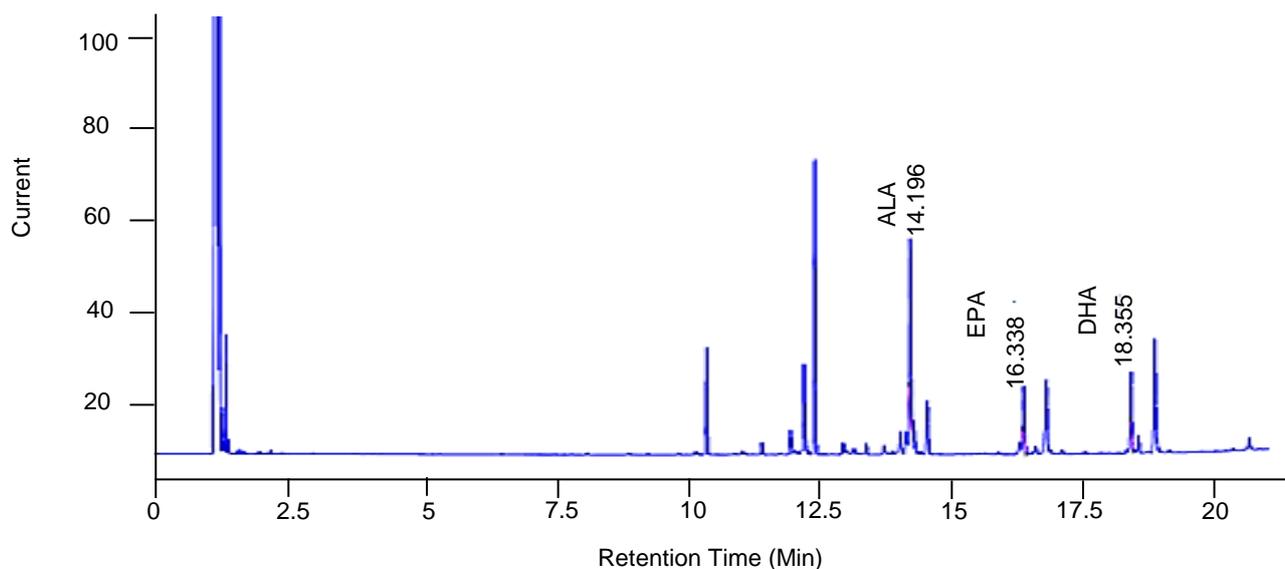


Figure 4. Representative chromatogram of lemuru fish oil fatty acids

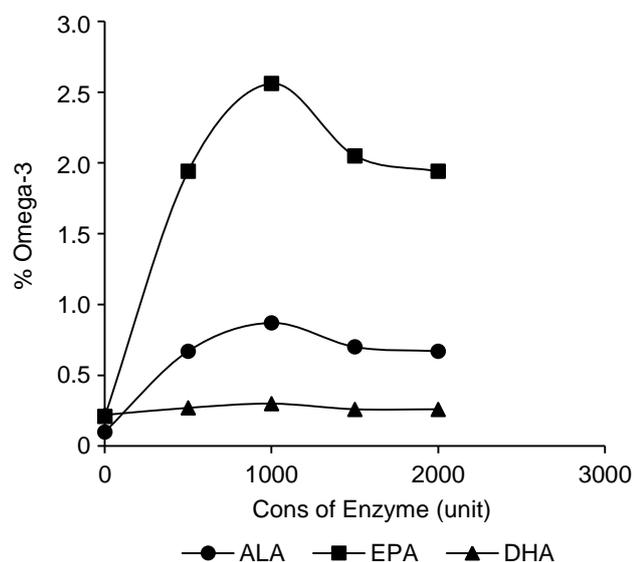


Figure 5. Effect of enzyme concentration on the omega 3 content of lemuru fish oil

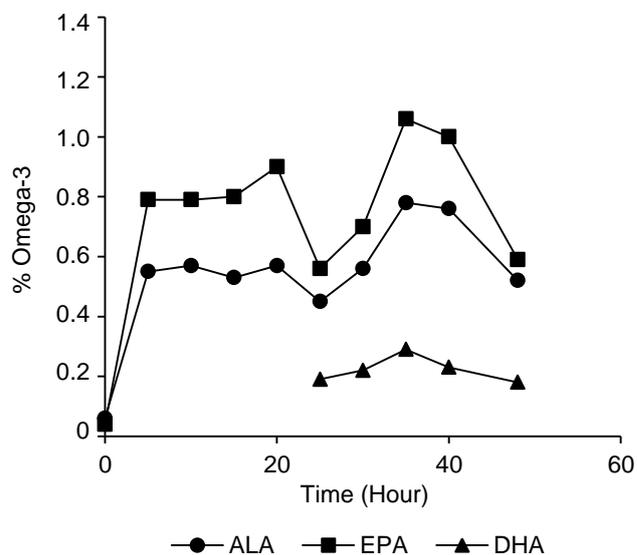


Figure 6. Effect of incubation time on the omega 3 content of lemuru fish oil

### The effect of incubation time on the omega-3 content

The enzymatic reaction time is a priority for products since a long enzymatic reaction might be detrimental. The enzymatic reaction must be controlled due to the reversible enzymatic reaction potential. The optimum incubation time was found by varying the incubation time to produce a high omega-3 content. Figure 6 explains that the highest ALA, EPA and DHA content was reached after 35 hours incubation time with a concentration of 0.78, 1.06, and 0.29% respectively, which shows a faster analysis outcome compared to the incubation time of 40 hours found in the previous study (Wanasundara and Shahidi, 1998).

### CONCLUSION

The results obtained in this study show that the combination of 3% activated carbon and 3% bentonite as an adsorbent allows for the purification of lemuru fish oil to improve its quality. The proposed procedure is rapid, easy and inexpensive, decreasing the FFA content and PV, thus improving the clarity of the lemuru fish oil. GC analysis also shows that the oil with the highest omega-3 (ALA, EPA and DHA) content was obtained using hydrolysis with lipase at a concentration of 1000 units for 35 hours. As the result, the omega-3 fatty acid content of lemuru fish oil can be increased tenfold.

This technique provides an accurate result and can improve the quality of lemuru fish oil. The results of the analysis provide a tolerance value that is within the standard limit of IFOS.

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